

# Numericals Chemistry Chapter Solid State

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and volume, rigidity, high density, low compressibility. The constituent particles (atoms, molecules or ions) are closely packed and held together by strong interparticle forces

**Types of Solids**

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Answer: (i) Those solids which have 3-D arrangement of atoms, ions or molecules are called crystalline solids, e.g. diamond. (ii) This is due to shifting of an ion from its normal position to an interstitial, in the crystal lattice, thus causing a vacancy in the original position. The overall density is not

affected by this change.

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Chemistry Notes for class 12 Chapter 1 The Solid State Solids Solids are the chemical substances which are characterised by define shape and volume, rigidity, high density, low compressibility. The constituent particles (atoms, molecules or ions) are closely packed and held together by strong interparticle forces

**Types of Solids**

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**Numericals Chemistry Chapter Solid State**

**Numericals Chemistry Chapter Solid State**

In the chapter Solid state Numericals are of four types - (a) Calculation of Rank of the unit cell (b) Relation between Edge length & Nearest neighbor distance (c) Edge length & radius (d) Density of unit cell.

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NCERT Solutions for Class 12 Chemistry Chapter 1 - The Solid State Solids deal with arrangements of particles in solids that result in several types of structure in solids. These differences in the structural unit also result in different properties of each Solid-State.

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Ans: The main concepts under this chapter are: Characteristics and properties of the solid-state Amorphous and Crystalline solids Classification of crystalline solids Crystal Lattices and unit cells Unit cells Types of unit cells Crystal lattices Characteristics of a crystal lattice Number of atoms ...

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Combination of elements of groups 12 and 16 yield some solid

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