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IN 10 MIN (no equipment) 5 Problem Areas
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30 minute workout)

Quick Burn

SADDLEBAGS

Slimdown! Best Outer

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Chest, \u0026
Shoulders Workout -
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Statistical Mechanics

Lecture 1 **Bestätigt**

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10 MIN (no

equipment) 5

Problem Areas

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kinetic energy of a

particle in two

dimensions is $K = p^2$

$2m = p^2 x + p^2 y$ 2m.

(1) By the Classical Equipartition Theorem, the mean kinetic

energy per particle is

$K = \frac{p^2}{2m} = \frac{p^2 x + p^2 y}{2m}$

$2m = 2p^2 x \cdot 2m = kT.$

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kinetic energy of a particle in two dimensions is $K = \frac{p_x^2}{2m} + \frac{p_y^2}{2m}$.

(1) By the Classical Equipartition Theorem, the mean kinetic energy per particle is $K = \frac{p_x^2}{2m} + \frac{p_y^2}{2m} = 2 \cdot \frac{1}{2} kT = kT$.

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 Problem Set #7. Due
 by 5 PM Friday, April
 11 th, 2003. Reif,
 Problems 7.1, 7.3, 7.5,
 7.7, 7.14, * 7.20.
 Problem 7: * The
 energy levels of a
 particle trapped in a
 cubical box of volume
 $V = L^3$ are given by $E_{n_x, n_y, n_z} = \frac{h^2}{8mL^2}(n_x^2 + n_y^2 + n_z^2)$. We can
 use this to model a
 single atom of helium
 trapped in a metal
 crystal. Phys416/517:
 Problem Set #7 The
 first part of this
 problem is the solution
 to Reif 7.24, shown
 here Problem 7.24 Reif
 The conditions of the
 problem make it
 reasonable to treat the
 loss of molecules from
 the vessel as
 effusion. We can also
 assume the reverse flux
 from outside the
 container is
 negligible. Let $N(t)$ be

the number of
 molecules remain-
 ...Problem 7.24
 Reif Problem 7.21 Reif
 We found in Problem
 7.20b that the
 translational kinetic
 energy probability
 density $\psi(\mathbf{r})$ for a gas
 with a Maxwellian
 speed distribution was
 given by the function
 $\psi(\mathbf{r}) = \frac{1}{\sqrt{2\pi\mu}} \frac{1}{\sqrt{2\pi\mu}} \frac{1}{\sqrt{2\pi\mu}} \exp[-\frac{m(v_x^2 + v_y^2 + v_z^2)}{2kT}]$. (1) The
 most probable energy
 e is the one that
 maximizes the value of
 $\psi(\mathbf{r})$. Thus, Problem 7.21
 Reif No need to sweat
 it, though, we have
 some potential
 solutions to the pesky
 iPhone 7 issues for all
 the Apple lovers who
 were forgotten. 1)
 Activation issues When
 activating a new
 iPhone 7,
 connect... Common
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 Solutions $T = -T \frac{\partial}{\partial T} \ln \mu \frac{\partial V}{\partial T}$. p. p., (7)

which is one form of the answer. The problem did ask for an answer in terms of the thermal expansion coefficient α , defined as $\alpha = \frac{1}{V} \left(\frac{\partial V}{\partial T} \right)_p$. p. (8)

After substituting Eq.(8) into Eq.(7), we have $\mu \partial C_p \cdot \partial p$. Reif Problem 5-13

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Reif There are several ways to calculate the entropy differences that are asked for. One way is to use the formulas found in Ch. 5, specifically Eq.(5.4.4), for the change in entropy of ideal gases. Another is to use ideal gas expressions for S that are developed in Ch. 7 based on statistical mechanics,

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use this to model a
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energy of a particle in
two dimensions is $K=$

$p^2/2m = p_x^2/2m + p_y^2/2m$. (1) By the Classical
Equipartition Theorem,
the mean kinetic
energy per particle is
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(2) This is
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Problem 7.21 Reif We
found in Problem 7.20b

that the translational kinetic energy probability density $\psi(\mathbf{v})$ for a gas with a Maxwellian speed distribution was given by the function $\psi(\mathbf{v}) = \left(\frac{m}{2\pi kT}\right)^{3/2} \exp\left[-\frac{m\mathbf{v}^2}{2kT}\right]$. (1) The most probable energy e is the one that maximizes the value of $\psi(\mathbf{v})$. Thus, *Solutions Manual for Introduction to Statistical Physics ...*

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expansion coefficient α , defined as $\alpha = \frac{1}{V} \frac{\partial V}{\partial T}$ ¶. p. (8) After substituting Eq.(8) into Eq.(7), we have $\mu = \frac{\partial C_p}{\partial p}$ ¶.
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Reif Problem 5-13

The first part of this problem is the solution to Reif 7.24, shown

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