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conventions. The mole The mole, abbreviated mol, is the SI base unit used to measure the amount of a substance. A mole is defined as the number ... except those required for the answer. Conversion: 3.5 dozen roses ...

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6.022x10²³particles((atoms,(molecules))

2. 1mole(=6.022x10²³particles((

1mole=molar(mass(1mole=22.4L(3.

Calculate(the ...

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saramukherjee. Terms in this set (32) one

mole is = to how many particles. 6.02 x 10

to the 23rd. abbreviation for a mole is.

mol. mole is the number of representative

carbon 12 particles in how many grams of

carbon 12.

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